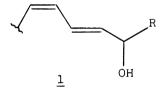
A New Stereoselective Synthesis of (E,Z)-Conjugated Hydroxy-Dienes, Key Intermediates for the Synthesis of HETES

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A general and highly stereoselective synthesis of (E,Z)-conjugated hydroxy-dienes is described and its synthetic utility is illustrated by the synthesis of precursors of ( $\pm$ )-LTA $_{\mu}$  methyl ester and ( $\pm$ )-5-HETE respectively.

Recently a new major pathway has been discovered for arachidonic acid metabolism which involves the conversion of this acid into the monohydroxyeicosatetranoic acids (HETES) by the action of lipoxygenases .  $^{1)}$  These monohydroxylated metabolites possess important biological properties and have been the subject of active investigations. It has been pointed out that all the HETES have a common structural moiety  $\underline{l}$  involving a (E,Z)-conjugated diene and a hydroxy group adjacent to the E double bond.



Two general approaches to these hydroxy-dienes have been recently described: the stereoselectivity of the first approach  $^2$ ) is not very high (EZ/ZZ = 75-80/25-20) and in the second one  $^3$ ) the Z double bond is generated via a Wittig reaction, the stereoselectivity being not specified. We report in this note a highly stereo selective synthesis of hydroxy-dienes  $\underline{l}$  and its application to the synthesis of precursors of  $(\pm)$ -LTA $_{ll}$  and  $(\pm)$ -5-HETE.

The starting point of our synthesis (Scheme 1) is the lactone  $\underline{2}$  easily available either in racemic  $\underline{4}$  or in optically active form .  $\underline{5}$ 

A one pot reduction (DIBAL, toluene, -78 °C) and Wittig-Horner olefination ((EtO)<sub>2</sub>POCH<sub>2</sub>CO<sub>2</sub>Et, n-BuLi, -78 °C RT, 15 h) led to the pure trans ester  $\underline{3}$  in 40-60% yield. After protection of the hydroxy group (dihydropyran, ether, p-TsOH, 75%) the ester  $\underline{4}$  was reduced (DIBAL, toluene, -78 °C, 80%) and the resulting alcohol was oxidized (MnO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 4 h, RT, 95%) to give the aldehyde  $\underline{5}$ . Nucleophilic addition of Grignard reagents (RMgBr, ether, 0 °C) provided alcohols  $\underline{6}$  (70-90%) as mixtures of two diastereoisomers in practically equal proportions.  $\underline{7}$ ) After silylation (TBDPSCl, DMF,

1928 Chemistry Letters, 1988

imidazole, 80%) the Z double bond was generated by a smooth retro-Diels-Alder reaction (xylene, 140 °C, 3 h, 80-85%) to provide (E,Z)-dienes  $\underline{8}$ . Deprotection of the primary alcohol (PPTS, ethanol, 55 °C, 3 h, 60%) afforded the (E,Z)-conjugated dienes  $\underline{9}$ , stereoisomerically pure as shown by  $^{1}$ H NMR .  $^{8}$ ) The allylic alcohols  $\underline{9}$  can be then converted to the corresponding bromides and coupled with cuprates of acetylenic compounds to give dienes of type  $\underline{1}$ . The viability of this sequence is illustrated by the synthesis of (2E,4Z,7Z)-tridecatrienyl tetrahydropyranyl ether ( $\underline{15}$ ) and of methyl 10-hydroxy-5-(t-butyldiphenylsilyl)oxy-(6E,8Z)-decadienoate ( $\underline{19}$ ), key intermediates for the synthesis of ( $\underline{\pm}$ )-LTA $_{h}$  methyl ester and ( $\underline{\pm}$ )-5-HETE respectively.

(2E,4Z,7Z)-tridecatrienyl tetrahydropyranyl ether  $(\underline{15})$  <sup>9)</sup> was obtained from the bicyclic ester  $\underline{3}$  (Scheme 2). After silylation of the primary alcohol (TBDPSCl, DMF, Imidazole, 93%) the ester group was reduced (DIBAL, toluene, 79%) and the created hydroxy group was protected (dihydropyran, pTsOH, 83%).

$$\frac{3}{10} \qquad \frac{11}{12} \text{ X=OTBDPS}$$

$$\frac{11}{12} \text{ X=OTBDPS}$$

$$\frac{11}{12} \text{ X=OTBDPS}$$

$$\frac{12}{12} \text{ X=OH} \qquad \frac{13}{13} \text{ X=Br}$$

$$\frac{CH_2OTHP}{C_5H_{11}} \qquad \frac{CH_2OTHP}{C_5H_{11}} \qquad \frac{(\pm)-LTA_4}{C_5H_{11}}$$

$$\frac{14}{14} \qquad \text{Scheme 2.} \qquad \frac{15}{15}$$

Chemistry Letters, 1988

The protected diol  $\underline{10}$  was then thermolyzed (xylene, 140 °C, 1 h) to give the (E,Z) diene  $\underline{11}$  (stereoisomeric purity> 95% as shown by  $^1$ H NMR). Desilylation (nBu $_{\mu}$ NF, THF) and conversion to the bromide (CBr $_{\mu}$ , ( $\Phi_2$ PCH $_2$ ) $_2$ , CH $_2$ Cl $_2$ ) afforded the bromodiene  $\underline{13}$  in 61% yield over the three steps. Coupling of  $\underline{13}$  with an excess of the cuprate of 1-heptyne (5 equiv. 1-heptyne, 5 equiv. C $_2$ H $_5$ MgBr, 0.5 equiv. CuCl in THF; the freshly prepared bromide  $\underline{13}$  was added at room temperature to this solution and the mixture was heated 1 h at 60 °C) gave  $\underline{14}$  in 65-71% yield. Semi-hydrogenation  $\underline{10}$  of  $\underline{14}$  (H $_2$ , Lindlar, hexane containing 2% quinoline by volume, 1.5 h, RT) provided the triene  $\underline{15}$  in 78% yield . $\underline{11}$  The synthesis of ( $\underline{\pm}$ )-LTA $_{\mu}$  methyl ester via the phosphonate derived from  $\underline{15}$  has been described by North . $\underline{9a}$ 

Methyl 10-hydroxy-5-(t-butyldiphenylsilyl)oxy-(6E,8Z)-decadienoate  $(\underline{19})^{-12}$  was obtained following Scheme 3. Thermolysis of aldehyde  $\underline{5}$  (xylene, 140 °C, 3 h) afforded the (E,Z)-diene  $\underline{16}$  in 87% yield. Addition of the lithio OBO orthoester derived from 1-(3-bromopropyl)-4-methyl-2,6,7-trioxabicyclo[2.2.2]octane  $^{13}$ ) to the aldehyde  $\underline{16}$  (1.2 equiv. of bromo orthoester, 2.4 equiv. t-BuLi -78 °C, 15 min, followed by addition of  $\underline{16}$ , -78 °C, 30 min) gave the dienol  $\underline{17}$  in 36% yield .  $^{14}$ ) Hydrolysis  $^{15}$  of  $\underline{17}$ , (AcOH, THF, H<sub>2</sub>O (4:2:1), 1.5 h, RT) followed by transesterification (K<sub>2</sub>CO<sub>3</sub>, MeOH) furnished the methyl ester  $\underline{18}$  in 66% yield for the two steps.

5 CHO
OTHP
OTHP
OTHP

$$16$$
 $17$ 
 $(CH_2)_3^{CO_2Me}$ 
 $(CH_2)_3^{CO_2Me}$ 

Scheme 3.

Protection of the secondary hydroxy group (TBDPSC1, DMF, imidazole) and deprotection of the primary alcohol (PPTS, ethanol, 55 °C) led to the (E,Z)-diene  $\underline{19}$  as a unique stereoisomer as shown by H<sup>l</sup> NMR .  $^{16}$ ) The synthesis of (±)-5-HETE by coupling of the bromide derived from  $\underline{19}$  with the cuprate of 1,4-decadiyne followed by semi-hydrogenation and hydrolysis has been established by Rokach .  $^{12}$ )

1930 Chemistry Letters, 1988

## References

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- 6) A Wittig-Horner olefination of the isolated lactol led always to a mixture of ester  $\underline{3}$  and of a tricyclic compound arising from an intramolecular Michael addition: see R. Bloch and M. Seck, Tetrahedron Lett., 28, 5819 (1987).
- 7) The transposition of this sequence to the synthesis of optically active  $\underline{1}$  (starting from an enantiomer of  $\underline{2}$ ) would demand a stereoselective addition of organometallic species to  $\underline{7}$ . Various solvents (ether, THF, HMPA), organometallics (RMgBr, RLi, RTi(OiPr)3) and hydroxyl protective groups (THP, TBDPS, MEM) have been tried but without any useful improvement of the selectivity: only small ratio changes from 50/50 to 60/40 have been observed by HPLC.
- 8)  $^{1}$ H NMR (250 MHz, CDCl<sub>3</sub>) for  $\underline{9}$  (R = C<sub>5</sub>H<sub>11</sub>) :  $\delta$  7.3 7.7 (m, 10H), 6.1 (dd, J = 11.4, 16 Hz, 1H), 5.95 (dd, J = 11.4, 10.1 Hz, 1H), 5.6 (dd, J = 6, 16 Hz, 1H), 5.4 (dt, J = 10.1, 6.3 Hz, 1H), 4.15 (m, 1H), 4.1 (d, J = 6.3 Hz, 2H), 1.1 1.6 (m, 9H), 1.05 (s, 9H), 0.9 (t, J = 7 Hz, 3H).
- 9) For previous syntheses of (2E,4Z,7Z)-tridecatrienol see a) J.C. Buck, F. Ellis, and P.C. North, Tetrahedron Lett., 23, 4161 (1982); b) S. Tsuboi, T. Masuda, and A. Takeda, Chem. Lett., 1983, 1829.
- 10) Over reduction of the triene occurred to the extent of 5-20% depending on the run. The best catalyst found was the commercial Lindlar purchased from Fluka.
- 11)  $^{1}$ H NMR (250 MHz, CDCl<sub>3</sub>) for  $_{15}$ :  $\delta$  6.64 (dd, J = 11.2, 15.2 Hz, 1H), 6.08 (dd, J = 11.2, 11 Hz, 1H), 5.85 (dt, J = 15.2, 6.2 Hz, 1H), 5.45 (m, 3H), 4.75 (m, 1H), 4.35 (m, 1H), 4.10 (m, 1H), 3.95 (m, 1H), 3.58 (m, 1H), 3.0 (dd, J = 6.2, 6 Hz, 2H), 2.1 (m, 2H), 1.3 1.9 (m, 12H), 0.95 (t, J = 7 Hz, 3H).
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- 13) E.J. Corey and N. Raju, Tetrahedron Lett., <u>24</u>, 5571 (1983).
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- 15) P.Y. Kwok, F.W. Muellner, C.K. Chen, and J. Fried, J. Am. Chem. Soc.,  $\underline{109}$ , 3684 (1987).
- 16) <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) for  $\underline{19}:\delta$  7.3 7.7 (m, 10H), 6.1 (dd, J = 11.3, 15 Hz, 1H), 5.95 (dd, J = 11.3, 11 Hz, 1H), 5.65 (dd, J = 15, 7 Hz, 1H), 5.5 (dt, J = 11, 7 Hz, 1H), 4.25 (m, 1H), 4.15 (d, J = 7 Hz, 2H), 3.6 (s, 3H), 2.2 (t, J = 7 Hz, 2H), 1.4 1.7 (m, 4H), 1.1 (s, 9H).

( Received August 11, 1988 )